

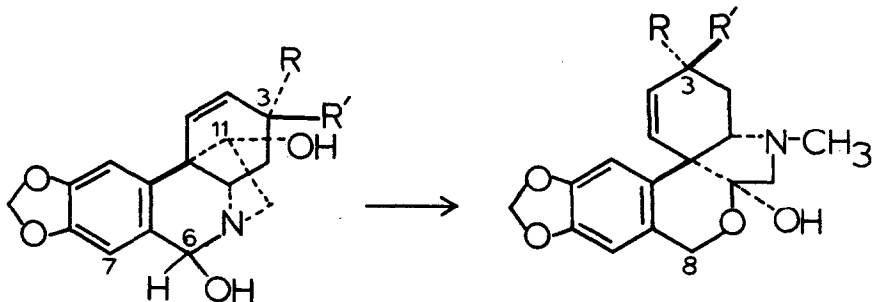
THE REARRANGEMENT OF 6-HYDROXYCRINAMINE TO CRIWELLINE

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The epimeric alkaloids haemanthidine (Ia) and 6-hydroxy-
crinamine (Ib) are known to undergo rearrangement to give the
corresponding C_3 -epimers tazettine (IIa) and criwelline (IIb)
respectively, upon treatment with methyl iodide followed by
dilute base (1-3). The C_3 -position seems not to be involved
in the rearrangement, and it is most probable that both con-
versions follow the same mechanistic pathway. An intramole-
cular hydride shift has been suggested for the rearrangement,
but no evidence for its validity has been presented to date (4).
The present communication provides evidence for the correct-
ness of our original mechanism.

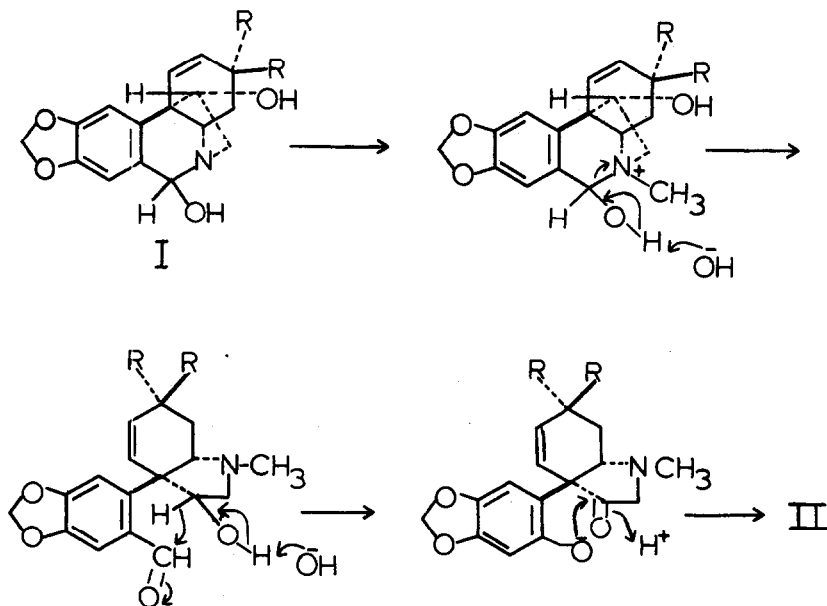


Ia) Haemanthidine; R=H, R'=OCH₃

IIa) Tazettine; R=H, R'=OCH₃

Ib) 6-Hydroxycrinamine; R=OCH₃, R'=H

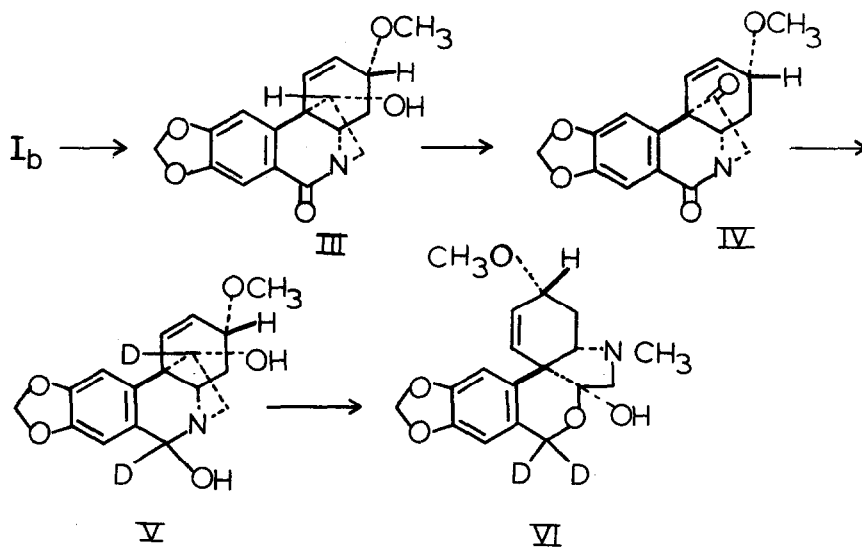
IIb) Criwelline; R=OCH₃, R'=H



Oxidation of 6-hydroxycrinamine (Ib) with manganese dioxide gave the bridgehead lactam (III),* m.p. 208-209°; $\lambda_{\text{max}}^{\text{EtOH}}$ 232, 274 and 323 μ ($\log \epsilon = 4.45, 3.88$ and 3.76 respectively). Reported (5): m.p. 195-196°; $\lambda_{\text{max}}^{\text{EtOH}}$ 234, 275 and 326 μ ($\log \epsilon = 4.36, 3.79$ and 3.71 respectively). Oxidation of III with chromium trioxide in dimethylformamide (6) containing a trace of sulfuric acid afforded IV, m.p. 199-200°; $\lambda_{\text{max}}^{\text{EtOH}}$ 224, 234, 277 and 326 μ ($\log \epsilon = 4.20, 4.21, 3.78$ and 3.56 respectively); $\lambda_{\text{max}}^{\text{KBr}}$ 5.72, 5.92 and 6.18 μ . The ketolactam (IV) was

* Satisfactory analyses have been obtained for all compounds cited in this communication.

reduced with lithium aluminum deuteride in tetrahydrofuran to 6-hydroxycrinamine-6,11-d₂ (V), m.p. 208-209°. Found: D, 1.59 atoms (7). Methylation of V with methyl iodide in acetone afforded a methiodide which was rearranged by dilute base to crivelline-8-d₂ (VI), m.p. 206-207°. Found: D, 1.74 atoms.



The nuclear magnetic resonance spectra of 6-hydroxycrinamine, crivelline and their deuterated analogs are given in figures 1 and 2. Proton assignments were made on the basis of expected chemical shifts and by comparison with other derivatives. Deuteration experiments showed that the C₆ and C₁₁ hydroxyl protons of Ib are centered at 8.0 and 2.4 p.p.m. respectively. The hydroxyl proton of crivelline appears at 2.9

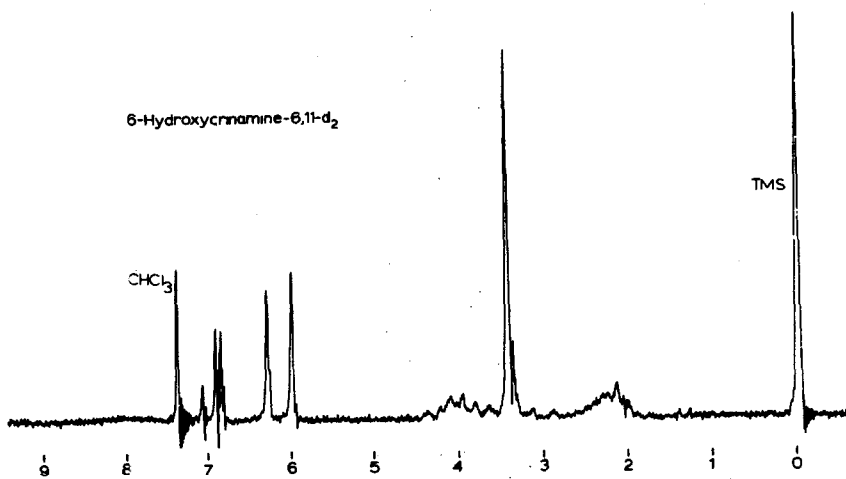
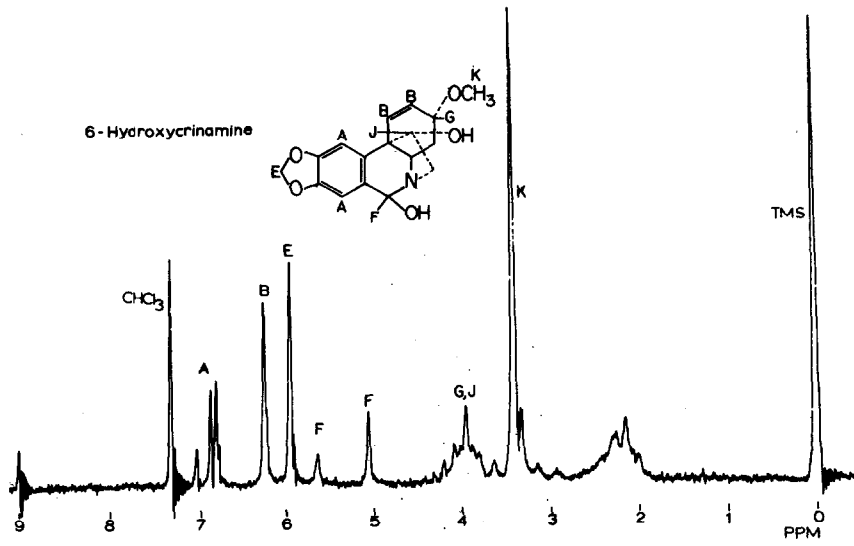


FIG. 1

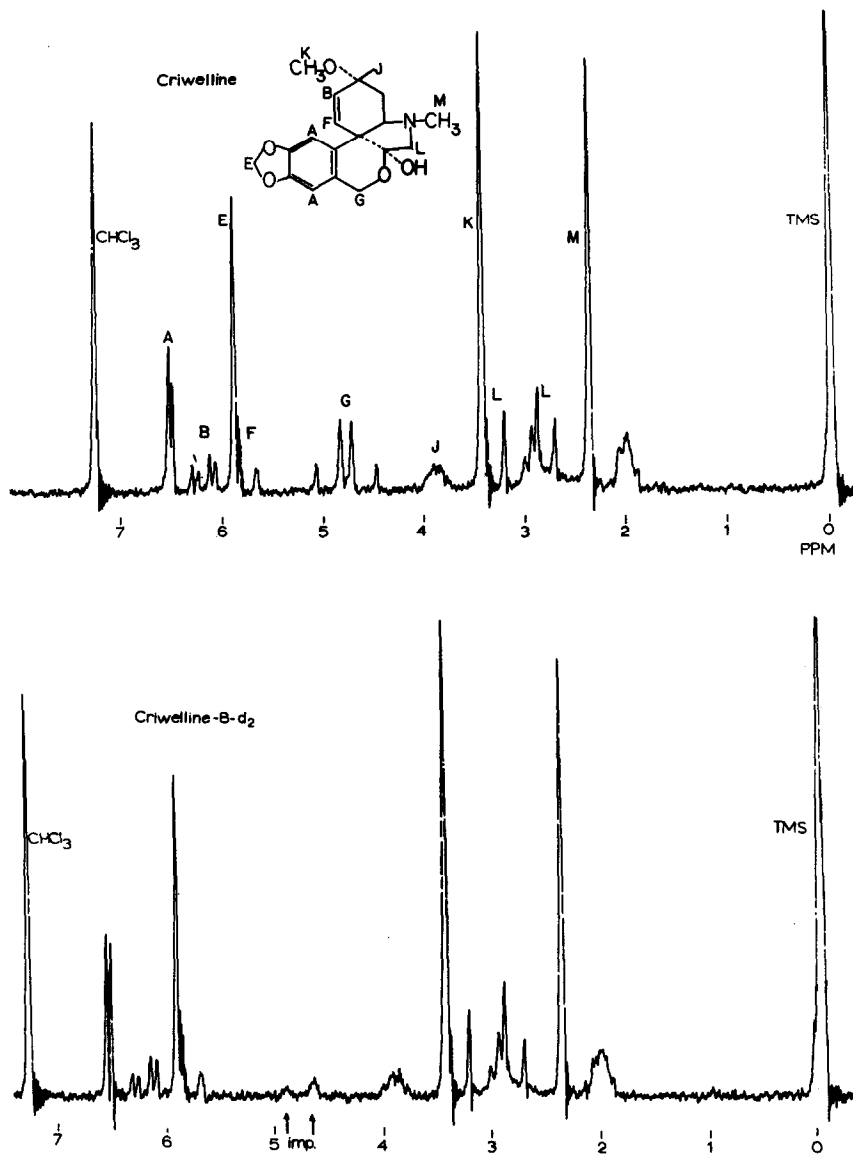


FIG. 2

p.p.m. The spectrum of 6-hydroxycrinamine is particularly significant because both the benzylic and C₇ protons appear at two different positions in the spectrum. The two peaks at 5.05 and 5.65 p.p.m. correspond to the benzylic proton (F), while peaks from 6.8 to 7.1 p.p.m. represent the aromatic protons. Concurrent spectroscopic studies of haemanthidine, haemanthamine and crinamine suggest that this phenomenon is due to a solution equilibrium of C₆ hydroxyl epimers and not to an impurity or a mixture of compounds.

The spectrum of criwelline shows an ABX splitting pattern ($J_{BF} = 10.6$ c.p.s., $J_{BJ} = 3.2$ and $J_{FJ} = 1.0$ c.p.s.) and two AB patterns ($J_{GG} = 15.5$ c.p.s., $J_{LL} = 11.5$ c.p.s.). The latter splittings arise because of the rigid ring system of criwelline. The spectrum of 6-hydroxycrinamine-6,11-d₂ is identical with that of the parent alkaloid except for the absence of the C₆ and C₁₁ proton resonances. The C₇ proton resonance is still in two positions because the equilibrium as related to the C₇ proton still exists. The criwelline-8-d₂ spectrum shows no AB pattern corresponding to resonances in the benzylic position. The absence of any benzylic proton resonance shows that two deuterium atoms now are substituted in that position.

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